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Chapter 3

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Chapter 3: Atoms 3-9.
(e) Ar is in Period 3, the
sixth element in the
block where the
orbitals are being
filled. There are 8
valence electrons in Ar,
found in the 3s and
3p orbitals. (f) Bi is in
Period 6, the third
element in the block

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where the orbitals are
being filled.

Chapter 3 Review, pages 186-191

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Section 3.1 questions.
p. 166 Question 10 ...

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3 Quiz. True/False.
Indicate whether the
sentence or statement
is true or false. T F. 1.
The region in space
where an electron is

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most likely to be found is called an energy level. T F. 2. Electron configurations are often condensed by writing them using the previous noble gas core as a starting point.

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Early Atomic Theories
and the Origins of
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Section 3.1 P. 173

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Below are all of the
resources for chapter 7
and 8. This is an
important unit because
there are a lot of
questions on the exam
and there are a lot of
labs in this unit. ... 7.1
p. 420 in the Nelson

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$(3.5 \text{ kg})(5.4 \text{ m/s}) + (4.8 \text{ kg})v_f = (3.5 \text{ kg})v_{f1} + (4.8 \text{ kg})v_{f2}$
 $18.9 \text{ m/s} + (4.8 \text{ kg})v_f = (3.5 \text{ kg})v_{f1} + (4.8 \text{ kg})v_{f2}$

The conservation of kinetic energy equation can be simplified by multiplying both sides of the equation by 2

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and noting that $v_i = 0$ m/s.

Answers
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3(s) with the energy
term separate is: 2
 $Al(s) + 3/2 O_2(g) \rightarrow Al$
 $2O_3(s) \Delta H_f^\circ = -1676$

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kJ/mol A

Chapter 5 Review Answers

thermochemical equation for the formation of aluminum oxide, $\text{Al}_2\text{O}_3(\text{s})$ with the energy term embedded in the equation is: $2 \text{Al}(\text{s}) + \frac{3}{2} \text{O}_2(\text{g}) \rightarrow \text{Al}_2\text{O}_3(\text{s}) + 1676 \text{ kJ}$

29. (a) Answers may vary. Sample answer: Hydrogen gas has a high enthalpy of ...

**Chapter 5 Review,
pages 338-339**

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Models and Properties

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1 Quiz. True/False.

Indicate whether the sentence or statement is true or false. 1. An ester is formed when the hydroxyl group of an alcohol and the hydrogen atom of the carboxyl group of an acid are eliminated, and water is released.

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2. Benzene is generally more reactive than alkanes and less reactive than alkenes

...

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Chapter 3 the
StruCture of Matter
and the CheMiCal
eleMentS 75 One
doesn't discover new
lands without
consenting to lose
sight of the shore for a

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very long time. Andre
Gide French Novelist
and Essayist n this
chapter, we begin the
journey that will lead
you to an
understanding of
chemistry.

Chapter t S of M and CheMiCal eleMentS

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Chemistry 7.1 The

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Nature of Solutions 10
Using Table 7.11, list
three salts containing
the sulphide anion that
would have a low
solubility in water at
25°C Many possible
answers For example:
FeS, CuS and ZnS 11

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