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QUESTIONS  $\text{CH}_4(\text{g}) + 2 \text{Cl}_2(\text{g}) \rightarrow$

$\text{CH}_2\text{Cl}_2(\text{g}) + 2 \text{HCl}(\text{g})$  3. Methane gas

reacts with chlorine gas to form

dichloromethane and hydrogen chloride,

as represented by the equation above.

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(a) A 25.0 g sample of methane gas is placed in a reaction vessel containing 2.58 mol of  $\text{Cl}_2(\text{g})$ .

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2009 edited May 2009 in AP Tests Preparation. Seriously??!!! I didn't learn how to do a couple of the questions themselves and what was up with the plethora of conceptual questions rather than math? Gah, I hope it's curved.

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